

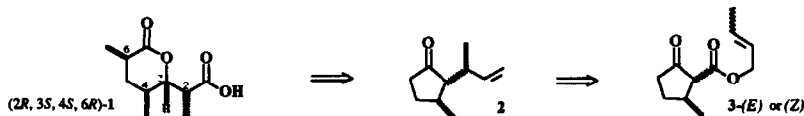
Stereoselective Ester Dienolate Carroll Rearrangement: a New Approach to the Prelog-Djerassi Lactone Framework

Nicole Ouvrard, Jean Rodriguez* and Maurice Santelli

Laboratoire de Synthèse Organique, Faculté des Sciences de St Jérôme, Boîte D12, 13397 Marseille cedex 13 - France

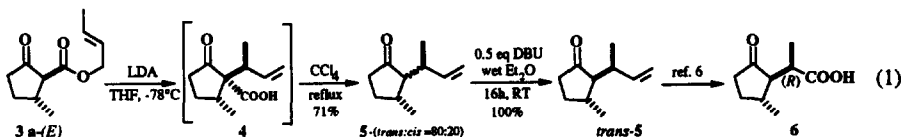
Abstract: The stereoselective ester dienolate Carroll rearrangement of (*E*) and (*Z*)-allylic β -ketoesters has been studied and found to be a new attractive approach to the synthesis of the Prelog-Djerassi lactone and related compounds. The synthesis of the optically active lactone (-)-(2*R*, 3*R*, 4*R*, 6*S*)-1 as well as the formal synthesis of its natural occurring isomer are described.

The Prelog-Djerassi lactone is an important synthon for the preparation of several macrolide antibiotics as it constitutes a key degradation product of various aglycons like methynolide and narbonolide.¹ Since the first preparation by Masamune in 1975, considerable attention has been given to the preparation of 1. Our own interest for the stereoselective synthesis of 1² is based on the preparation of a substituted γ,δ -ethylenic ketone 2 as a key intermediate, which could be formed by the stereoselective Carroll rearrangement^{3a-c} of an allylic β -ketoester 3.

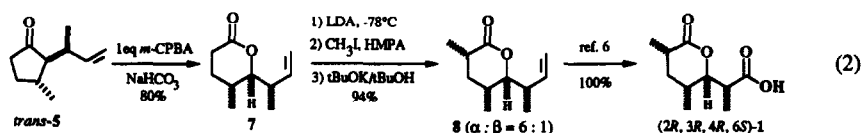


Although a number of recent works have demonstrated the synthetic potentiality of [3,3] sigmatropic rearrangements,^{3f} only few reports on the stereoselective Carroll rearrangement have been published recently.^{3d,e} Herein, we wish to report our preliminary results on the stereoselective synthesis of (2*R*, 3*R*, 4*R*, 6*S*)-1 by using the ester dienolate Carroll rearrangement of the (*E*)- β -ketoester 3a as a new efficient approach to the Prelog-Djerassi lactone and related compounds. The optically active (2*S*, 3*R*)- β -ketoesters 3a and 3b serve as model compounds for our study and are easily prepared by transesterification⁴ of the corresponding methyl ester² in quantitative yield. The correct (2*R*, 3*S*) enantiomer can be formed in good yield from (*S*)-(-)-pulegone.²

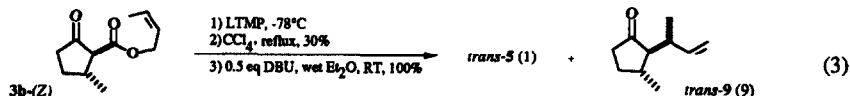
Treatment of 3a (45.0 mmol, 8.8 g) with 2.5 equiv of LDA in THF at -78 °C followed by decarboxylation^{3b} of the crude rearranged β -ketoacid 4 leads to the γ,δ -ethylenic ketone 5 as a mixture of *trans* and *cis* isomers in the ratio 80/20, which upon epimerization gives rise to pure *trans*-5 in 71% overall yield.⁵ The configuration of *trans*-5 was confirmed by an X-Ray structural determination of the ketoacid 6 obtained by oxidative cleavage of the double bond (eq 1).⁶



The synthesis was completed by the regio- and chemoselective Baeyer-Villiger oxidation of 5 (8.3 mmol, 1.26 g) to the δ -lactone 7 which was alkylated and subjected to thermodynamically equilibrated conditions to give predominantly 8 bearing α -Me with a ratio of α : β =6:1 easily purified by medium pressure chromatography on silica gel (76% overall).⁷ RuO₄ oxidation⁶ yielded quantitatively (2*R*, 3*R*, 4*R*, 6*S*)-1⁸ (eq 2).



The synthesis of unnatural (2*S*, 3*R*, 4*R*, 6*S*)-1 was similarly attempted starting from the (*Z*)-(2*S*, 3*R*)- β -ketoester 3b. However, in this case, the rearrangement of the dilithium dienolate (LDA, -78 °C) proceeds with only moderate diastereoselectivity at the newly created asymmetric center (c.a. 40%). An improved selectivity has been obtained with lithium 2,2,6,6-tetramethylpiperidine (LTMP) at -78 °C leading after decarboxylation and epimerization to a mixture of *trans*-5 and *trans*-9 in the ratio 1/9 with 30% overall yield providing a formal stereoselective synthesis of natural occurring 1 (eq 3).



The synthetic sequence based on the stereoselective ester dienolate Carroll rearrangement giving rise to (2*R*, 3*R*, 4*R*, 6*S*)-1 associated to the diastereoselective preparation of the key intermediate 9 constitutes a new interesting approach to the synthesis of 1 and related compounds. Works aimed at the completion of the synthesis of the natural occurring Prelog-Djerassi lactone as well as the study of the scope and limitation of this approach are under active investigations.

Acknowledgment. We gratefully acknowledge the generous support of Rhône-Poulenc-Rorer (Vitry) and we are particularly grateful to Dr. J. C. Barrière (Rhône-Poulenc-Rorer) for helpful discussions.

References and notes

- For a recent review, see : Martin, S.F.; Guinn, D.E. *Synthesis*, **1991**, 245-262.
- For previous works from our laboratory, see : Hacini, S.; Santelli, M. *Tetrahedron*, **1990**, *46*, 7787-7792.
- (a) Carroll, M.F. *J. Chem. Soc.*, **1941**, 507-511. (b) Wilson, S.R.; Price, M.F. *J. Org. Chem.*, **1984**, *49*, 722-725. (c) Snider, B. B.; Beal, R.B. *J. Org. Chem.*, **1988**, *53*, 4508-4515. (d) Gilbert, J.C.; Kelly, T.A. *Tetrahedron*, **1988**, *44*, 7587-7600. (e) Echavarren, A.M.; de Mendoza, J.; Prados, P.; Zapata A. *Tetrahedron Lett.*, **1991**, *32*, 6421-6424. (f) Ziegler, F.E. *Chem. Rev.*, **1988**, *88*, 1423-1452.
- Gilbert, J. C.; Kelly, T.A. *J. Org. Chem.*, **1988**, *53*, 449-450.
- trans*-5: colorless oil, $R_f = 0.42$, ether/pentane: 1/9; $[\alpha]_{578}^{22} = +128.8$ (CHCl₃, $c = 2$); IR (CCl₄) 3080, 1745, 1640 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) $\delta = 0.98$ (d, $J = 7.1$ Hz, 3H), 1.02 (d, $J = 6.0$ Hz, 3H), 1.26 (m, 1H), 1.63 (dm, $J = 10.3$ Hz, 1H), 1.96 (m, 3H), 2.19 (dd, $J = 16.4, 7.1$ Hz, 1H), 2.59 (m, 1H), 4.89 (d, $J = 10.0$, Hz, 1H), 4.92 (d, $J = 17.1$, 1H), 5.71 (ddd, $J = 17.1, 10.0, 7.3$ Hz, 1H); ¹³C-NMR $\delta = 16.28, 20.24, 29.58, 33.68, 36.42, 38.77, 60.88, 114.00, 141.69, 219.74$; Anal. Calcd. for C₁₀H₁₆O: C, 78.95; H, 10.53. Found: C, 78.80; H, 10.63.
- Carlsen, H.J.; Katsuki, T.; Martin, V.S.; Sharpless, K.B. *J. Org. Chem.*, **1981**, *46*, 3936-3938.
- Suzuki, K.; Tomooka, K.; Katayama, E.; Matsumoto, T.; Tsuchihashi, G. *J. Am. Chem. Soc.*, **1986**, *108*, 5221-5229.
- (2*S*, 4*R*, 5*R*, 6*R*)-1: greenish crystals, m.p. = 90 °C; $R_f = 0.32$; ether; $[\alpha]_{578}^{26} = -12.86$ ($c = 1.5$ CHCl₃); IR (CCl₄) 3300-3180 (broad), 1740, 1720, 1240, 1010 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) $\delta = 0.97$ (d, $J = 6.6$ Hz, 3H), 1.23 (d, $J = 7.1$ Hz, 3H), 1.29 (d, $J = 7.2$ Hz, 3H), 1.30 (q, $J = 13.0$ Hz, 1H), 1.87 (ddd, $J = 3.4, 6.0, 13.4$ Hz, 1H), 2.16 (m, 1H), 2.51 (dq, $J = 6.0, 13.0$ Hz, 1H), 2.84 (dq, $J = 2.6, 7.2$ Hz, 1H), 4.10 (dd, $J = 2.6, 10.4$ Hz, 1H), 8.80 (m, 1H); ¹³C-NMR $\delta = 13.19, 17.10, 17.41, 31.41, 36.07, 37.28, 41.94, 87.77, 174.51, 177.95$; HRMS calcd. for C₁₀H₁₆O₄: 200.1048. Found: 200.1047.